

Ultra-bright Designer Photocathodes

Katherine Harkay
Argonne, APS

Future Light Sources Workshop
JLab, Newport News, VA

March 8, 2012

Acknowledgements

Argonne:

Károly Németh

Richard Rosenberg

Badri Shyam

George Srajer

Joseph Zsolt Terdik (student)

Marion White

Zikri Yusof

Sasha Zholents

Illinois Institute of Tech (IIT):

Linda Spentzouris

Jeff Terry

Aditya Unni

Chris Kalnmals (student)

Daniel Velazquez (student)

LBNL:

Gennadi Lebedev

Thanks to: *D. Dowell, I. Bazarov, C. Hernandez-Garcia, W. Hess, R. Legg, B. Militsen, H. Padmore, T. Rao, J. Smedley, W. Wan*

Outline

- Introduction
- Approach, results, and plans:
 - Design
 - Synthesize
 - Characterize
- Summary



Introduction

- Ultra-bright photocathodes are a key technology for the development of future light sources.
- Two approaches are applicable, depending on time-frame:
 - **Near-term:** Optimize the synthesis and performance of long-known photocathodes such as Cs₂Te, CsKSb, GaAs (ANL, ASTeC, BNL, HFZD, INFN, JLab, LBNL, PITZ, et al.)
 - **Mid-to-far-term:** Explore novel crystal systems numerically and optimize ("design") their properties, or nano-engineer surfaces, or other novel idea (**ANL**, ASTeC, BNL, Eindhoven, Jlab, LBNL, SLAC, UCLA, Vanderbilt, et al.).
- Properties to design or tune include:
 - Intrinsic emittance
 - Workfunction
 - QE
 - Reliability, robustness (vacuum, E-field, field emission)

Approach to cathode design

- **Design** (*accelerator & condensed matter physics, computational chemistry*)
 - Derive cathode requirements from future light source requirements.
 - Investigate candidate systems; compute electronic properties.
 - Select systems with promising predicted properties for synthesis.
- **Synthesize** (*chemistry, materials science*)
 - Develop & test hypotheses for chemical reactions.
- **Characterize** (*surface science, materials science*)
 - Develop portable load-lock system.
 - Develop cross-compatible sample transfer system.
 - Compare measurements with predictions.



Design

Main tool:

- Quantum-chemical Density Functional Theory (DFT)-based surface-physics calculations.

Two codes used:

- PWSCF (crystal lattice parameters, workfunction, gap energy, surface energy)
- YAMBO (optical absorption spectra)

Two of our cathode concepts:

- Tuning workfunction and intrinsic emittance using ultrathin films.
- Tuning the workfunction of Cs_2Te .

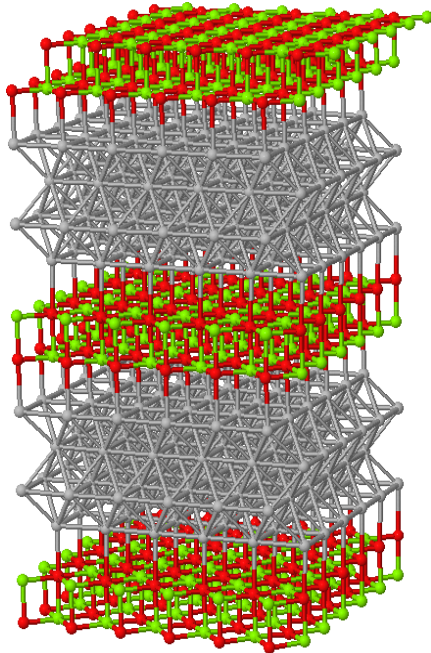


Potential low-transverse-emittance layered structure

Preliminary idea from surface catalysis systems [L. Giordano et al., J. Chem. Phys. 127, 144713 (2007)].

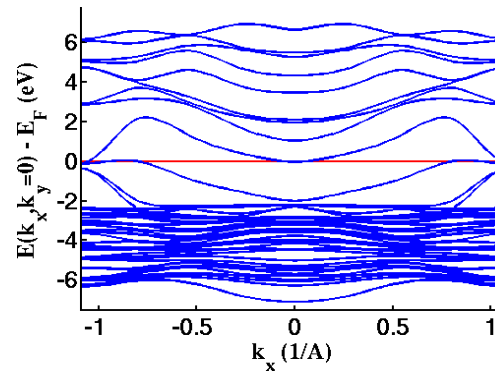
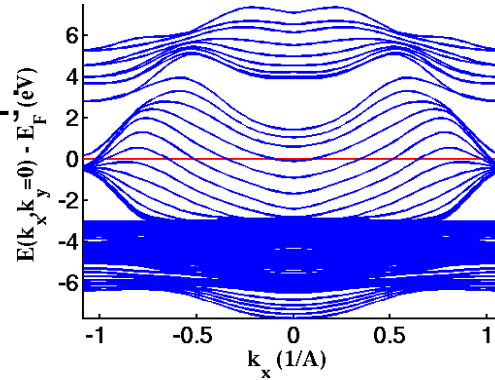
K. Nemeth et al., PRL 104, 046801 (2010).

Ag(001)16L
DFT(PBE)
 $k_{\max} 0.1 \text{ \AA}^{-1}$

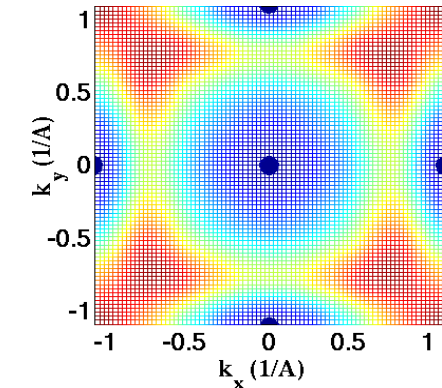
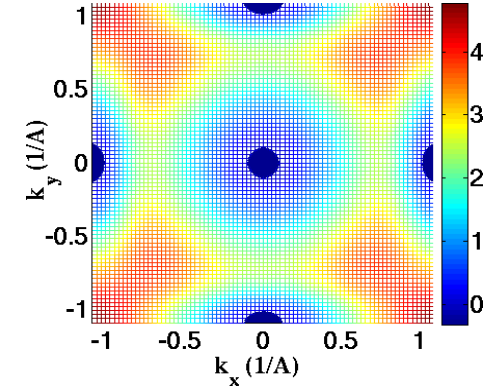


$k_{\max} 0.05 \text{ \AA}^{-1}$

k_x vs. energy bands relative to E_f



Lowest-energy (relative to E_f) surface bands in k_x vs. k_y space

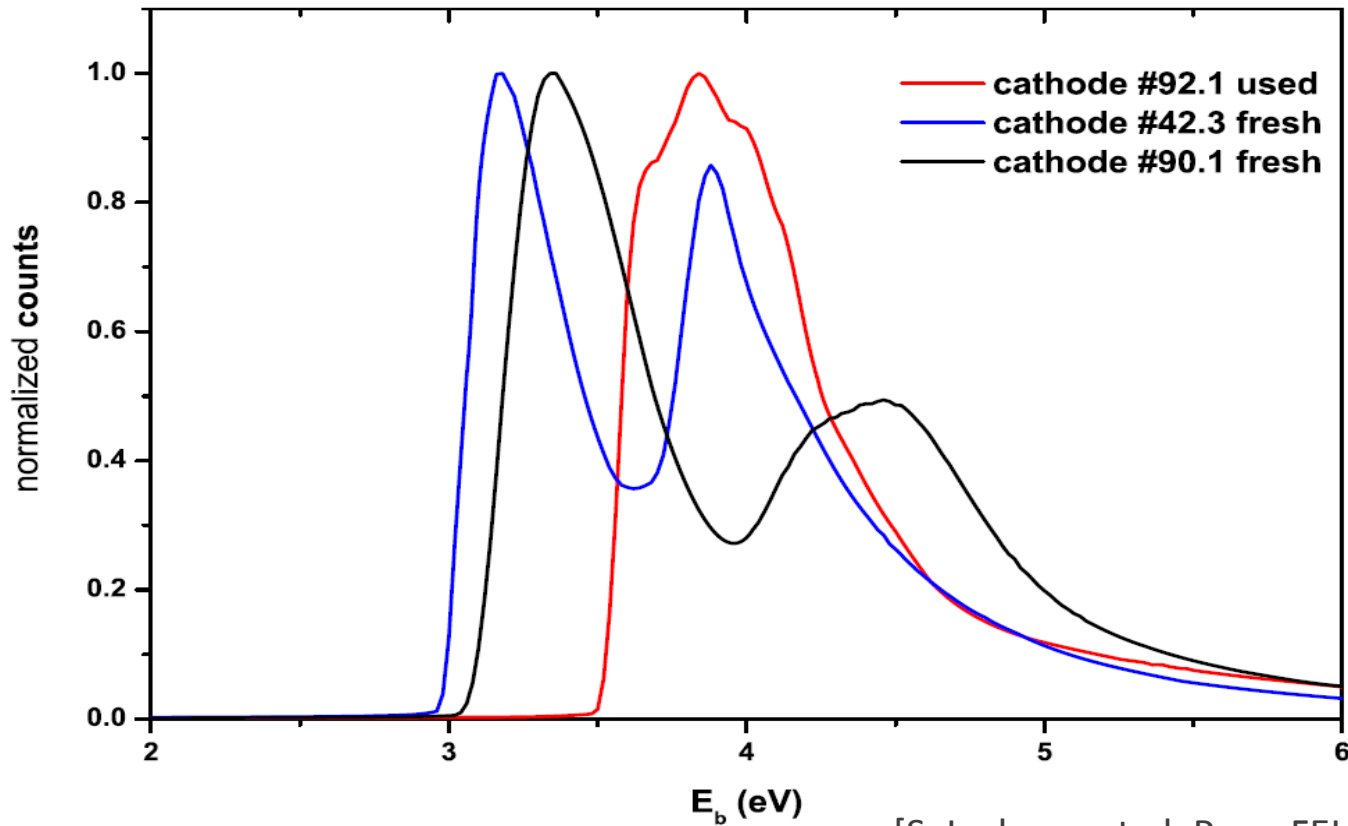


MgO(100)2L-Ag(100)4L-MgO(100)2L; DFT(PW91)
Work function reduced by ~ 1.6 eV relative to pure Ag(001)

$$\varepsilon_{n,i} / \sigma_x \quad 0.18 \text{ mm-mr/mm}$$

$$E_{\text{thermal}} \quad 30 \text{ meV}$$

Photoemission from Cs₂Te cathode



[S. Lederer, et.al, Proc. FEL 2007, p. 457]

- Workfunction increases and QE decreases as Cs₂Te ages.
- Pushing the Cs₂Te photoemission (workfunction) to visible/infrared would have tremendous advantages!

Quantum efficiency of Cs₂Te exposed to small gas molecules, such as O₂, CO₂, CO, N₂ and CH₄

[A. di Bona, et.al. J.Appl.Phys. 80 (5), 1996]

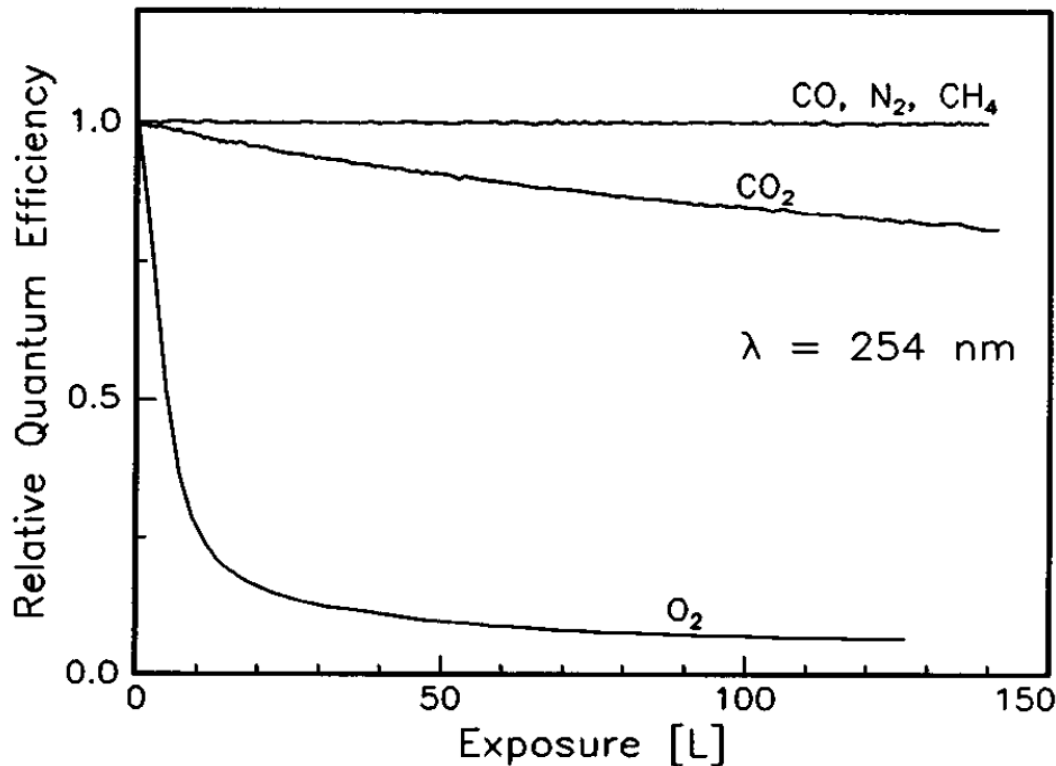


FIG. 7. Quantum efficiency drop for the 254 nm radiation of the photocathode exposed to different gases.

- Another important, relatively common small molecule, **acetylene**, C₂H₂ (e.g. for welding), was not investigated.
- **What about reacting Cs₂Te with acetylene?**
- Acetylene is fairly reactive, easily losing its hydrogens, which takes electrons away from the material in contact. This forms “acetylides,” containing the acetylide anion [C₂]²⁻, commonly denoted as C₂²⁻.

A_2MC_2 Type Ternary Acetylides

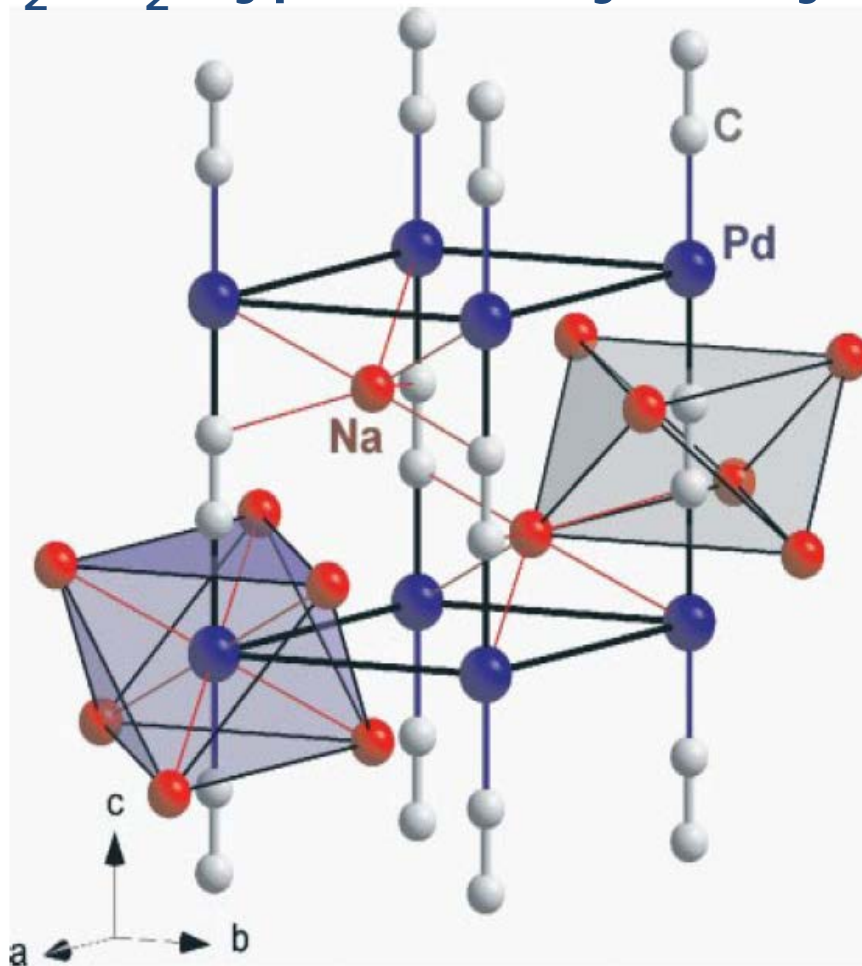


Fig. 3 Crystal structure of Na_2PdC_2 ($P\bar{3}m1$, $Z = 1$). Selected interatomic distances are: Pd–C 200.2(2) pm, Na–Pd 296.8(2) pm, Na–C 263.1(1) pm, C–C 126.3(3) pm.

[Fig: U. Ruschewitz, Z. Anorg. Allg. Chem. 632, 705, 2006.]

- Unique 1D substructures, where A=alkali metal and M=transition metal or metalloid.
- All known A_2MC_2 are colored semiconductors with 2.1-2.8 eV direct bandgaps.
- Synthesized so far: $M \in [Pt, Pd]$ and $A \in [Na, K, Rb, Cs]$.
- **We investigated acetylated Cs_2Te , where $M=Te$ and $A=Cs$**
- Completely new class of materials for photophysics and low-workfunction applications (e.g., field-emission displays).

Provisional patent application filed: K. Nemeth et al.

Experimental vs. DFT-calculated workfunctions (ϕ)

TABLE III: Experimental and calculated (DFT) properties of photoemissive surfaces of validation materials: workfunctions (Φ), bandgaps at the Γ -point $E_g(\Gamma)$ and surface energies (σ).

Compound and surface	Φ (eV)		$E_g(\Gamma)$ (eV)	σ (eV/Å ²)
	EXPT	DFT	DFT	DFT
Cs(100)	2.14 [21]	2.00	0.29	0.005
Te(001)	4.95 [21]	5.02	0.54	0.036
Cs ₂ Te(001)	2.90-3.0 [22]	3.08	0.77	0.015
Cs ₂ Te(010)	2.90-3.0 [22]	2.90	1.04	0.014
(Cs)Na ₃ KSb	1.55 [23]	-	-	-
K ₂ CsSb	1.9-2.1 [24, 25]	-	-	-

J.Z. Terdik, K. Nemeth, K. Harkay, et al, submitted for publication.

- Good agreement for Cs, Te, Cs₂Te; we are confident of predictions for theoretical systems with these components.
- Note low workfunctions for K₂CsSb and (Cs)Na₂KSb (engineered for high QE).

Calculated ϕ for selected A_2MC_2

- Significant ϕ anisotropy among rod-perpendicular (001) and rod-parallel, (110) and (010), orientations.
- Cs_2TeC_2 (010) and Cs_2PdC_2 (010) are photo-emissive in visible to near-IR!
- Comparable-to-better ϕ than K_2CsSb and $(Cs)Na_2KSb$ (1.55-2.1 eV).
- Rod-perpendicular surface could act like a 2D nanotip array.

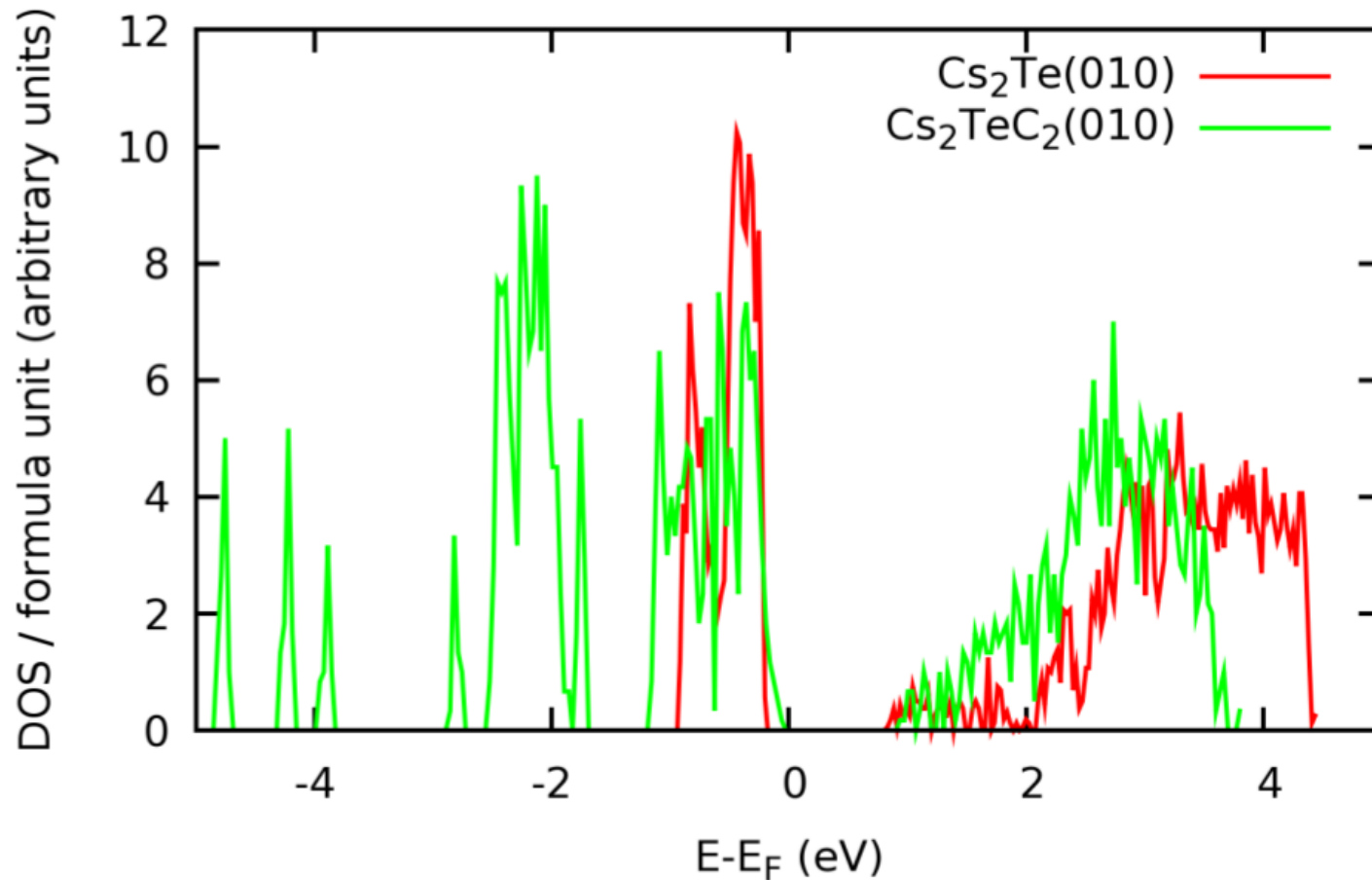
TABLE IV: Calculated (DFT) properties of photoemissive surfaces of acetylide compounds: workfunctions (Φ), bandgaps at the Γ -point $E_g(\Gamma)$ and surface energies (σ). Relaxed slabs refer to the relaxation of unrelaxed ones with the central 2 layers fixed. For h- $Cs_2C_2(001)$ and $Na_2TeC_2(010)$, $E_g(\Gamma) \approx 0.05$ eV has been found for a single band above E_F as well.

Compound and surface	unrelaxed			relaxed		
	Φ (eV)	$E_g(\Gamma)$ (eV)	σ (eV/Å ²)	Φ (eV)	$E_g(\Gamma)$ (eV)	σ (eV/Å ²)
o- $Cs_2C_2(010)$	2.80	1.25	0.023	-	-	-
h- $Cs_2C_2(001)$	2.56	1.14	0.027	-	-	-
$Na_2PdC_2(001)$	3.58	1.13	0.067	-	-	-
$Na_2PdC_2(110)$	3.73	1.65	0.029	4.17	2.34	0.024
$Na_2PdC_2(010)$	2.65	1.91	0.019	2.68	2.45	0.017
$Cs_2PdC_2(001)$	2.90	1.43	0.046	-	-	-
$Cs_2PdC_2(110)$	2.73	0.88	0.026	2.73	1.16	0.022
$Cs_2PdC_2(010)$	1.33	0.78	0.015	2.03	1.74	0.013
$Na_2TeC_2(001)$	3.40	1.03	0.029	-	-	-
$Na_2TeC_2(110)$	3.80	0.91	0.025	4.67	2.04	0.009
$Na_2TeC_2(010)$	2.75	1.43	0.015	2.68	1.34	0.015
$Cs_2TeC_2(001)$	3.71	1.86	0.022	-	-	-
$Cs_2TeC_2(110)$	2.77	0.77	0.020	2.98	1.38	0.019
$Cs_2TeC_2(010)$	1.71	1.00	0.013	2.44	1.63	0.009

J.Z. Terdik, K. Nemeth, K. Harkay, et al, submitted for publication.

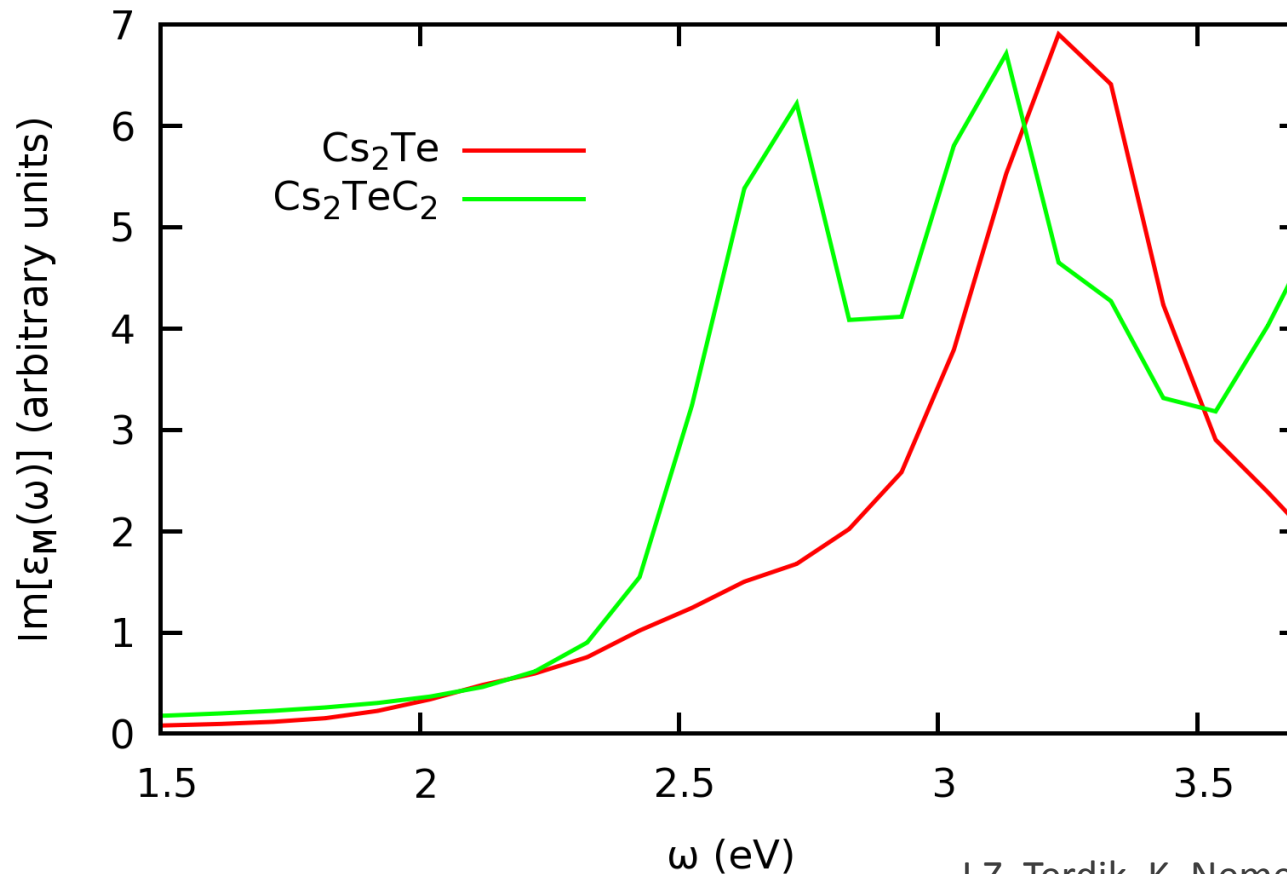
Density of states (DOS)

DOS of Cs_2TeC_2 is comparable to Cs_2Te ; computed using PWSCF (Terdik, Nemeth).



QE estimation

- Computed the optical absorption spectra (i.e., imaginary part of dielectric constant) using YAMBO (Nemeth).
- QE of Cs_2TeC_2 expected to be comparable to Cs_2Te .



J.Z. Terdik, K. Nemeth, K. Harkay, et al, submitted for publication.

Synthesis/Characterization

- Collaborations in place, but we welcome more!
 - PPNL working on MgO-Ag system (see W. Hess's talk)
 - IIT-ANL working on ternary acetylide systems. *Patent application in process, K. Nemeth et al.*
- Two modes are possible for synthesis/characterization:

	pros	cons
separated	Specialized expertise: consistent cathode quality/performance.	Load-lock sample transport needed.
combined in one system	No transport needed.	Long learning curve to achieve consistent quality/perform.

- Preference for option 1 (ternary acetylides):
 - Local Cs₂Te expert: Z. Yusof and colleagues (ANL/AWA) (QE 10%).
 - Expertise in synthesis and surface science and instrumentation is distributed (ANL/APS, ANL/ MSD, IIT, LBNL, etc.)

Characterization

Goal:

- Was desired compound synthesized? X-ray diffraction, XPS, ...
- Does it perform as predicted? Workfunction, QE, ARPES (intrinsic emittance), ...

Tools available or planned:

- Load lock system (vacuum hardware, sample-transfer hardware)
- Kelvin probe (workfunction)
- ARPES at LBNL/ALS beamline (being commissioned, G. Lebedev)
- ARPES at ANL, IIT (under development)

Formation of Na_2PdC_2 as observed by X-ray diffraction in the traditional synthesis. [U. Ruschewitz, Z. Anorg. Allg. Chem. 632,705 (2006)].

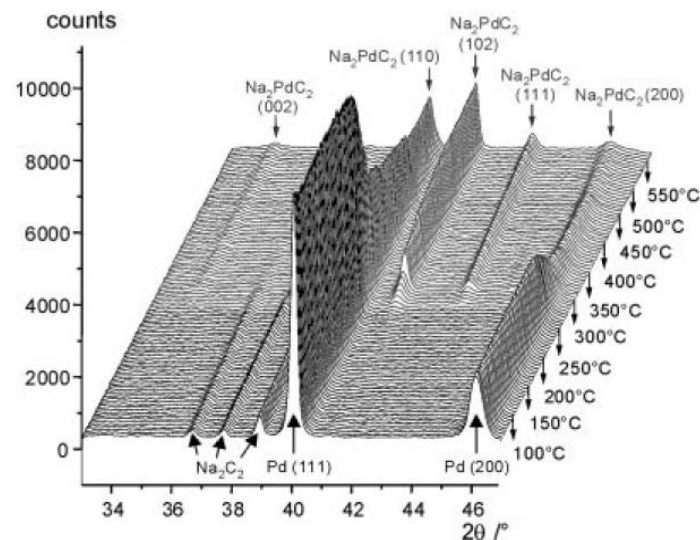


Fig. 1 Formation of Na_2PdC_2 from Na_2C_2 and palladium with increasing temperature as measured on an X-ray powder diffractometer (Huber G644, $\text{CuK}\alpha_1$ radiation). The reflections of the starting materials and the product are assigned.

Cs₂Te Photocathode - Fabrication and Vacuum Transfer

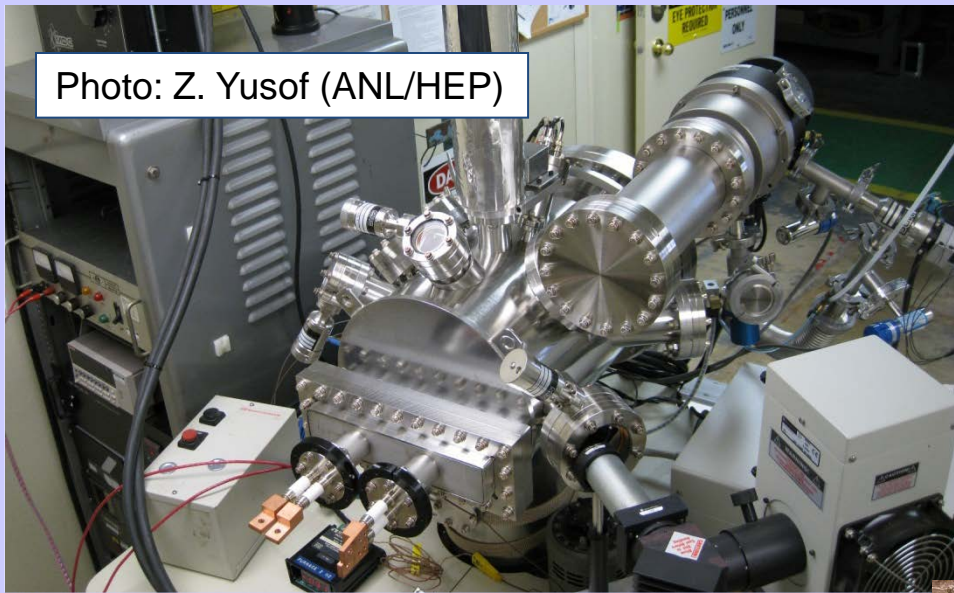
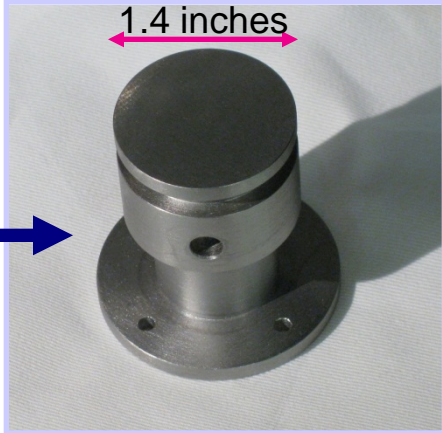


Photo: Z. Yusof (ANL/HEP)



Photocathode plug/substrate (Mo)

Photo: Z. Yusof

↑
Cs₂Te photocathode deposition system

translator

Portable UHV vacuum load-lock system →

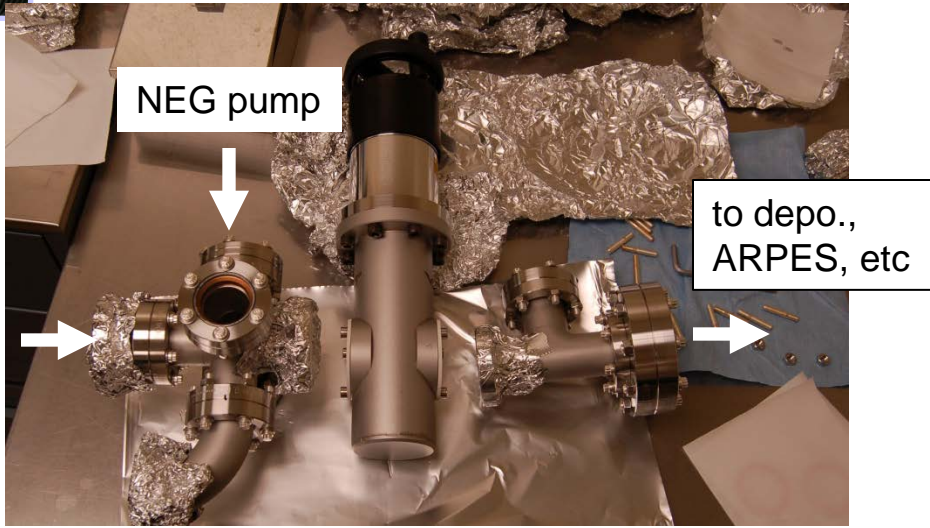
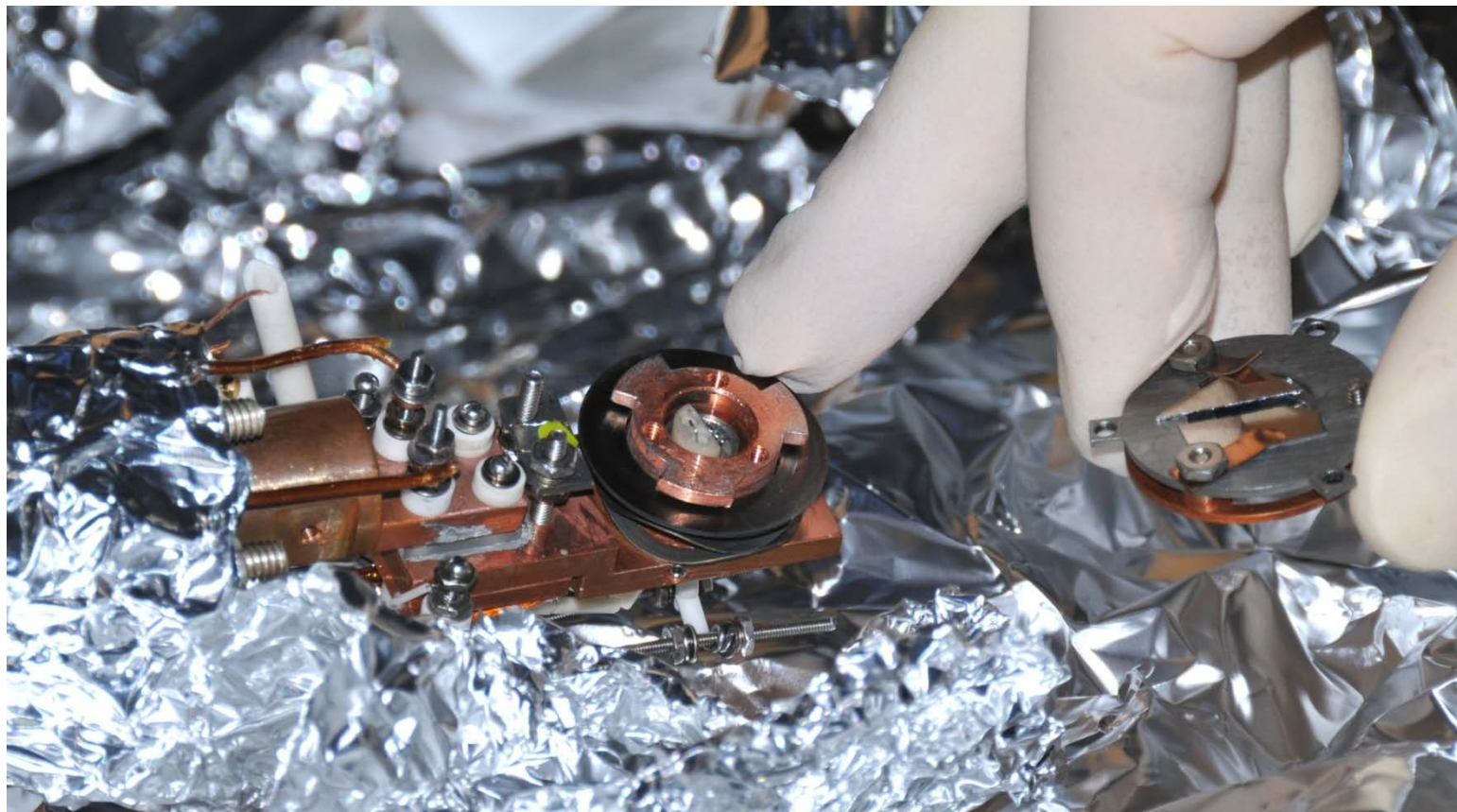


Photo: D. Velazquez (IIT)



Cross-compatible sample transfer hardware to be installed inside load-lock



Example: designed by R. Rosenberg.

FY12 Plans

1: Test transfer & characterization of Cs₂Te

- Transfer Cs₂Te from ANL to LBNL
- ALS sample transfer hardware to be adapted to AWA Cs₂Te chamber and load-lock system.
- Measure intrinsic emittance in ARPES TOF system being installed on ALS (G. Lebedev).

2: Ternary acetylide

- Synthesize Cs₂TeC₂ or other ternary acetylide at IIT and characterize at APS (QE, workfunction, structure).
- Synthesize/transfer Cs₂TeC₂ and characterize in LBNL and ANL or IIT ARPES systems.

Beyond FY12:

- Transfer to and characterize performance in an electron gun.
- Where? LBNL APEX VHF injector, Jlab, APS Injector Test Stand, ...

Resources

2009 Sept, DOE/BES Workshop on Accelerator Physics of Future Light Sources

D. Dowell et al., Nucl. Instrum. Meth A 622, 685 (2010).

Physics of Photocathodes for Photoinjectors

2010, Oct 12-14, BNL

<http://www.bnl.gov/pppworkshop/>

2012, Oct 8-10, Cornell

Announcement & web site posted soon

Several European photocathode workshops

Summary

- Ultra-bright photocathodes are a key technology for future light sources.
- We are thinking “outside the box” to tune photocathode properties to enhance future light source performance. More work is needed.
- MgO-Ag system has been shown experimentally to reduce the workfunction; calculations predict ultra-low transverse emittance.
- “Designer” acetylated $\text{Cs}_2\text{Te} - \text{Cs}_2\text{TeC}_2$ – and other systems in this class are predicted to have low workfunctions in the visible to near-IR range. The QE of Cs_2TeC_2 is predicted to be comparable to Cs_2Te . The rod-perpendicular orientation may exhibit low emittance.
- Synthesis of both ideas is ongoing through collaborations (PNNL and IIT/ANL, respectively) and additional collaborators are welcome.
- Characterization using surface science techniques is planned at APS and at LBNL/ALS beamline ARPES system; load-lock transfer needed.
- Characterization in a gun of promising cathodes is desired; load-lock transfer needed.